

-21-76 Hydrolysis of 100 mg ~~RNA~~<sup>DNA</sup> for For Pak I Isolation

- The ~~same~~ amount of DNA on one rod from JE-3 is  $\sim 10$  mg (previous page)
- take 10 rods -
- put 20 ml  $\text{HCOOH}$  (88%) into 50 ml beaker
- dissolve DNA from each stick into the  $\text{HCOOH}$  and let the beaker stand <sup>60 min</sup> with sticks in it and intermittent stirring - for one hour.
- evaporate off  $\text{HCOOH}$  under  $\text{N}_2$  -

Observation - after  $\sim 4$  hours there was still about 3-4 ml left - the solution was quite viscous. The sample spent  $\sim 5$  min at  $45^\circ\text{C}$  while I tried to rotovap it.

→ did not freeze

- sample stored in freezer overnight - at  $\sim 4$  ml level.
- freeze sample in dry ice-ethanol and try to lyophilize with this system:

The sample froze



- This system was good for keeping the sample in the solid state, but lyophilization proceeded very slowly
- sample was thawed, and kept at reduced P, but volume reduction again was slow.
- add 10 ml 10%  $\text{MeOH}$  to sample (which had a 2.5 ml volume)
- Cool in ice - lot of DNA started to ppt
- spin at 12k for 15 min in 40 rotor
- decant clear liquid, store in freezer.

## SOLUBILIZATION OF RADIOACTIVITY

50  $\mu$ l sampled out of 12.25 ml  $\rightarrow$  good recovery  
(actual original vol = 10 + 2.5 = 12.5 ml)

$$\text{DILUTION FACTOR} = \frac{12.5 \text{ ml}}{0.05 \text{ ml}} = 250$$

total counts in 50  $\mu$ l = 70,445 cpm  
total counts in 12.5 ml =  $1.76 \times 10^7$

from P 24, there are  $223,250 \text{ cpm/mg DNA}$   
 $\Rightarrow 2.233 \times 10^7 \text{ pm in original sample}$

$$\% \text{ Solubilization of Radioactivity} = \frac{1.76 \times 10^7}{2.233 \times 10^7} \times 10^2 = \underline{\underline{79\%}}$$

} could be adduct(s), degradation products, non-specifically or loosely-bound AFB, or metabolites.

## PERCENT LIBERATION OF ADDUCT FROM DNA

amount submitted to HPLC: 50  $\mu$ l (reps. 1/250 of sample)  
total cpm in two adduct peaks: 49,239

$$\Rightarrow (49239)(250) = 1.231 \times 10^7 \text{ cpm in these peaks in the entire sample}$$

% of original DNA radioactivity that is liberated as adducts : 55%

note that these hydrolyzed and were not opt for adduct lib.

% of soluble hydrolysate that is adducts : 70%

The rest of the sol. material may be AFB, degn. prod. or adduct still bound to soluble DNA.

Subject Peak 1 Isolation (cont) Instructor's Name

7-23-76

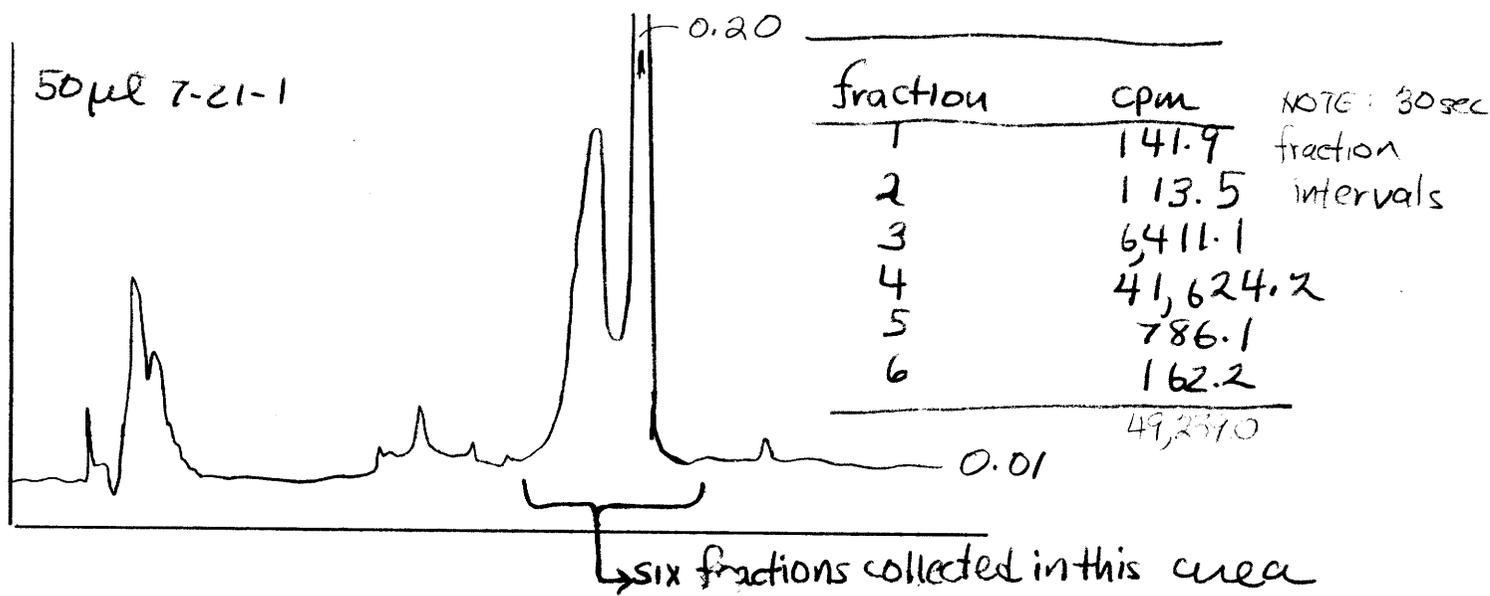
Sample 7-21-1 was thawed and prepared for analysis. It looked quite clean - cleaner than I remember it looking last night.  
- count 50  $\mu$ l

50  $\mu$ l 7-21-1 at the 12.25 ml level:  
281,788 counts/4 min = ~~70,445 cpm~~ <sup>70,445 cpm</sup>

- Inject 50  $\mu$ l into HPLC (waters column)

40 min gradient from 10% to 80% MeOH  
T=51, P=1130, v=1.0

Results: 79% of the radioactivity started with was in the liquid that was counted - the DNA (depurinated) largely was soluble in the 10% MeOH, however.



Subject Peak 1 Isolation (cont.) Instructor's Name

7-23-76 Counting "DNA" Pellet

A "DNA" pellet formed after centrifuging the cloudy material down that formed after addition of 10% MeOH. This was washed with MeOH, EtOH and then dissolved in  $H_2O : HCOOH$  (~1:2). 100  $\mu$ l of the ~~sol~~ solution was counted

100  $\mu$ l solubilized pellet: 143,644 c/10 min  
 $\Rightarrow 14,364.4$  cpm/0.1 ml  
total volume: ~10 ml (exact vol not meas)

$\Rightarrow 1.436 \times 10^6$  cpm in entire sample

~ THIS IS ~ 6% OF THE TOTAL CPM IN THE DNA ~  
which isn't really enough to worry about. (note, a little may have been lost in the isolation, but not enough to make a large difference).